

## Root Elongation, Root Surface Area and Organic Acid by Rice Seedling Under Al<sup>3+</sup> and/or H<sup>+</sup> Stress

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**Abstract: Problem statement:** Under acidic condition, Al<sup>3+</sup> is the most common species in solution. An experiment was conducted to study the effects of Al and/or pH on rice seed germination, root morphology and organic acids release. This study was conducted at ambient temperature in Malaysia. **Approach:** Two experiments were conducted: (1) Rice seeds undergoing germination were exposed to 0.5 mM CaCl<sub>2</sub> solutions containing various concentration of Al (10, 20, 30, 40-50 μM) and (2) The seeds were soaked in water taken from an acid sulfate soil area in Malaysia for which the pH was adjusted to a range of values using 0.01 M HCl or NaOH. **Results:** Root length decreased with increasing Al concentration, while the opposite was true for pH. The trend for the change of root surface area with Al concentration and pH is the same as that of root length. The critical Al concentration for rice growth is 15 μM. This means that rice variety MR 219 grown on 90% of the granary areas in Malaysia is relatively less tolerant compared to other rice varieties. At low pH and high Al concentration, the rice roots secreted citrate and/or oxalate which subsequently formed Al-citrate and Al-oxalate, respectively. This, to a certain extent, had reduced Al toxicity. This is the mechanism of rice tolerance to Al toxicity. **Conclusion:** Acid sulfate soils in Malaysia allocated for rice production should be limed to increase water pH in the paddy field to 5.0. Then, rice can grow without Al<sup>3+</sup> and/or H<sup>+</sup> stress.

**Key words:** Acid sulfate soil, organic acid, rice root length, acidic condition, paddy field, root elongation, root surface area

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### INTRODUCTION

Aluminum is an important component of the earth's crust, making it the third most abundant element after oxygen and silicon. Plant roots are, therefore, almost always exposed to Al in some form or other. Fortunately, most of this Al occurs as harmless oxides and alumino-silicates. However, when soils become acidic as a result of natural processes or due to human intervention, Al is solubilized into the toxic trivalent cation, Al<sup>3+</sup>. This form of Al is the most common species under acidic condition (Shamshuddin *et al.*, 2010). Kochian *et al.* (2005) found that in soils with pH value below 5, toxic forms of Al are solubilized into soil solution, inhibiting root growth. Aluminum toxicity has been recognized as a major threat to crop production in the tropics, such as corn (Shamshuddin *et al.*, 1991; Ismail *et al.*, 1993), cocoa (Shamshuddin *et al.*, 2004b) and oil palm (Auxtero and Shamshuddin, 1991) which are grown on moderately and very acidic soils in Malaysia.

Al<sup>3+</sup> is toxic to many plants even at micro-molar concentration (Shamshuddin *et al.*, 1991; Muhrizal *et al.*, 2003; Shamshuddin *et al.*, 2004a). A range of plant species has evolved mechanisms to enable them to grow satisfactorily on acidic soils that have toxic concentrations of Al<sup>3+</sup> which can hinder crop growth. Organic acids play a central role in these aluminum tolerance mechanisms. Some plants detoxify aluminum in the rhizosphere by releasing organic acids that chelate aluminum, rendering it unavailable to the growing crops. Citrate and malate can be present in the root tips. Root exudates organic acids to reduce the effects of Al toxicity, for example by forming Al-citrate or Al-malate. Among gramineous crops, rice (*Oryza sativa* L.) exhibits the highest level of Al-tolerance, despite of its quite low efflux capacity for organic acids under Al-stress (Ma *et al.*, 2002).

In Malaysia, acid sulfate soils which are dominated by pyrite (FeS<sub>2</sub>) are abundant, occurring almost exclusively along the coastal plains (Shamshuddin and Auxtero, 1991; Shamshuddin *et al.*, 1995; Muhrizal *et al.*, 2006; Enio Kang *et al.*, 2011). When the pyrite laden soils in the coastal plains are opened up for crop production or otherwise, high acidity (soil pH < 3.5) is

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produced, leading to release of high amount of Al into the soil environment (Shamshuddin *et al.*, 2004a), killing plants and aquatic life. Some of these soils are cropped to rice, variety MR 219, which is not known to be tolerant to Al toxicity. The yield of this rice variety grown on such acid soils is low, far below the national average of 3.80 t/ha. Because of this some of the paddy fields in the area are abandoned by the farmers (Enio Kang *et al.*, 2011). The acid sulfate soil infertility can be ameliorated by the application of lime, basalt, organic fertilizer and/or their combinations at appropriate rate (Suswanto *et al.*, 2007; Shazana *et al.*, 2011).

As the Malaysian government is now thinking of increasing self-sufficiency in rice consumption from 73 to 86%, areas under acid sulfate are targeted for future development, to grow rice. However, the pH of water in the paddy fields of acid sulfate soils is low (Shamshuddin and Auxtero, 1991; Auxtero and Shamshuddin, 1991; Shamshuddin *et al.*, 2004a) and Al concentration is above the known toxic level for rice growth of 74  $\mu\text{M}$  (Dent, 1986). The high Al concentration in the water from the paddy fields on acid sulfate soils would cause much problem to the farming community. The elongation of rice root is limited by the presence of excess amount of Al in the water.  $\text{Al}^{3+}$  is attracted to the root due to the negatively-charged cells wall. This phenomenon would affect rice growth and eventually the yield. The infertility of these soils needs to be alleviated using appropriate amendments.

For cultivation of rice in Malaysia, direct seeding is often practiced by the farmers. If the rice seeds come into contact with the toxic water in the paddy fields on acid sulfate soils, germination and growth of the rice will be affected. A study was, therefore, conducted to determine the effects of high Al concentration and low pH on seed germination, root morphology and organic acids release. This would hopefully help mitigate acid sulfate soil infertility for rice cultivation.

## MATERIALS AND METHODS

**Location, material and experimental:** This study was carried out at the Faculty of Agriculture, Universiti Putra Malaysia, Serdang, Malaysia. In this study two short-term experiments were conducted in the laboratory:

- This was a seed germination experiment conducted in a laboratory at the ambient temperature. Seeds of rice (variety MR 219) were soaked with a hormone-based chemical (Zappa™) for 24 h,

rinsed twice with distilled water so as to sterilize the outer cover of the seeds and then left in the dark place for 24 hours. This rice variety is commonly grown in Malaysia, planted on about 90% of the area allocated to rice. The pre-soaked seeds were then transferred into test tubes containing 0.5 mM  $\text{CaCl}_2$  solution with various concentrations of  $\text{AlCl}_3$  (10, 20, 30, 40, 50  $\mu\text{M}$ ). Every tube contained 3 seeds (seedlings)

- This was also seed germination experiment, except that it used acid water containing high concentration of Al taken from paddy field in Merbok, Kedah, Malaysia. The paddy field was located on a soil classified an acid sulfate soils (Sulfaquepts). This area was once under mangrove and was reclaimed for rice cultivation in 1964 (Chaang *et al.*, 1993). The pH of the water was determined by a pH meter, while Al concentration in the water was determined by atomic absorption spectrophotometer (AAS). The same variety of rice seeds was used in this experiment. The pre-soaked seeds were exposed to 100 mL of the acid water. For this experiment, the pH of the water was adjusted to various levels (3, 4, 5, 6-7) using 0.01 M HCl or 0.01 M NaOH. The number of seeds in the tube was also three

The experiments were laid out in Completely Randomized Design (CRD), replicated 4 times. The seeds were allowed to germinate and grow into seedlings for 7 days. No fertilizers were added to promote the growth as there would be enough nutrients in the seeds to have healthy growth of rice seedlings during the experimental period.

**Determination of root morphology:** At harvest, root length (cm) and root surface area ( $\text{cm}^2$ ) were measured using a root scanner, model Win RHIZO 2008a.

**Determination of organic acids in the root:** After 7 days, the rice plants were harvested from the tubes and the solution in each tube was filtered using 0.45  $\mu\text{m}$  Milipores® filter. These solutions were analyzed for the presence of organic acids by HPLC, using Aminex HPX-87H column. Filtrate samples (100  $\mu\text{L}$ ) were injected into the HPLC using a glass syringe and eluted isocratically with 0.008 N  $\text{H}_2\text{SO}_4$  at a constant flow rate of 0.6 ml/min for 25 min at 20°C. Peaks for the organic acids were detected at a wavelength of 210 nm and the organic acids were identified by comparing with the retention times obtained for pure organic acids injected as standards. From the peak areas, the quantity of organic acids in the samples was calculated and expressed as  $\mu\text{M}$ .

**Statistical analysis:** Data from the experiment were analyzed statistically using analysis of variance (ANOVA) and the Tukey's multiple range tests was employed to determine the significance of the differences between treatments. The statistical package used was SAS statistical software package (Version 9.1)

## RESULTS

**Chemical properties of water from the field:** The soils (acid sulfate soils) from where the acid water was taken for this experiment were classified as Sulfaquepts (Soil Survey Staff, 2010). For this study, we had taken water from a soil pits dug for profile description in order to characterize and classify the soils. Results from our analysis showed that the pH of the water was 3.70. The concentration of Al in the water was 878  $\mu\text{M}$ .

**Effects of Al and pH on root length:** In this study, it was observed that the root length of rice seedling was affected by the presence of high concentration of Al (Fig. 1). Root length is negatively and highly correlated with Al concentration (Fig. 1A). The relationship is presented by the equation  $Y = -1.33x + 17.92$  ( $R^2 = 0.83$ ).

Like the case of Al, root length of rice was affected by low pH. Figure 2 shows that rice root length is positively and highly correlated with pH. The equation representing the relationship is given by  $Y = 4.79x + 1.23$  ( $R^2 = 0.90$ ) (Fig. 2A).

Figure 3A and 3B show the relationship between relative root length and Al concentration and between relative root length and pH, respectively. Relative root length is negatively related with Al concentration, with equation representing the relationship given by  $Y = 171.88x^{-0.24}$  ( $R^2 = 0.82$ ) (Fig. 3a).

Relative root length of rice seedlings is positively correlated with pH (Fig. 3B). The relationship between relative root length and pH is given by the equation  $Y = 100.88 \ln(x) - 90.39$  ( $R^2 = 0.96$ ).

**Effects of Al and pH on root surface area:** Root surface area was determined in this study. It was observed root surface area of rice seedling was affected by Al (Fig. 1). The root surface area of rice seedling is negatively and highly correlated with Al concentration and the equation representing the relationship is given by  $Y = -0.11x + 1.45$  ( $R^2 = 0.95$ ) (Fig. 1B).

Root surface area of rice seedling increased with increasing pH (Fig. 2). Root surface area of rice seedling was positively and highly correlated with pH, with equation representing the relationship given by  $Y = 0.38x + 0.18$  ( $R^2 = 0.93$ ) (Fig. 2B).

**Exudation of organic acids:** Table 1 gives data on the exudation of organic acids as affected by Al. It shows that the amount of oxalic acid released by the 3 seedlings was not significantly different among the treatments. Nevertheless, it shows that higher oxalic acid was released at 50  $\mu\text{M}$  of Al concentration. The trend in the release of citric and malic acids is the same as that of the oxalic acid. Rice seedlings grown under the condition of Al concentration of 50  $\mu\text{M}$  had released about 0.1300, 0.0806-0.0766  $\mu\text{M}$ , oxalic, citric and malic acids, respectively.

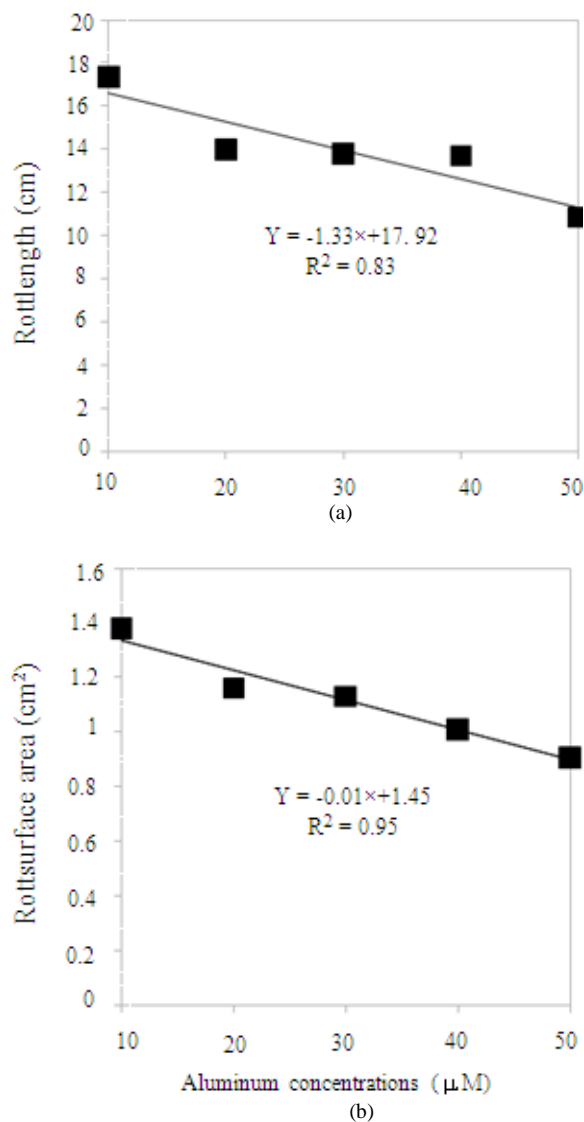


Fig. 1: Relationship between root length and Al concentration (A) and root surface area and Al concentration (B)

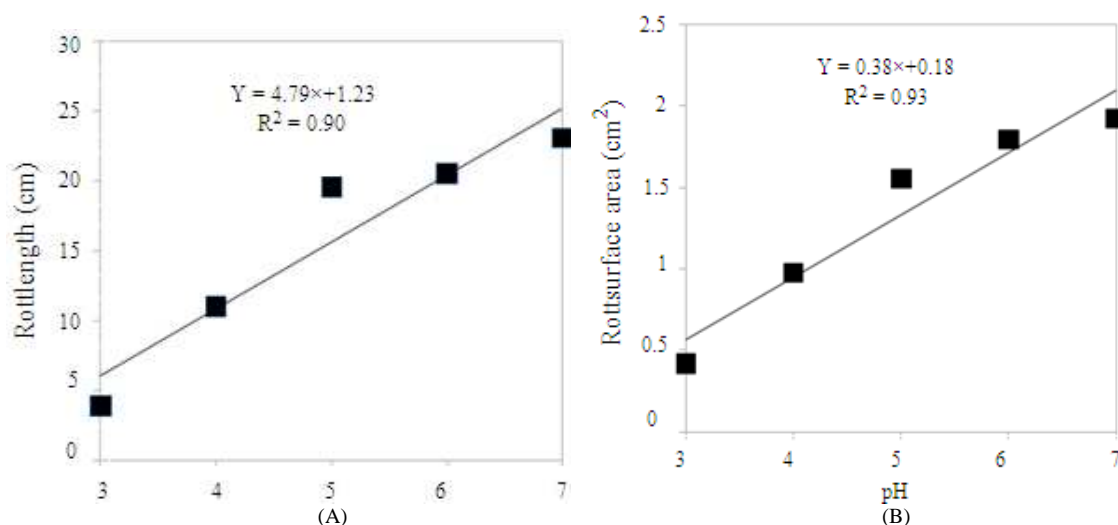


Fig. 2: Relationship between root length and pH (A) and root surface area and pH (B)

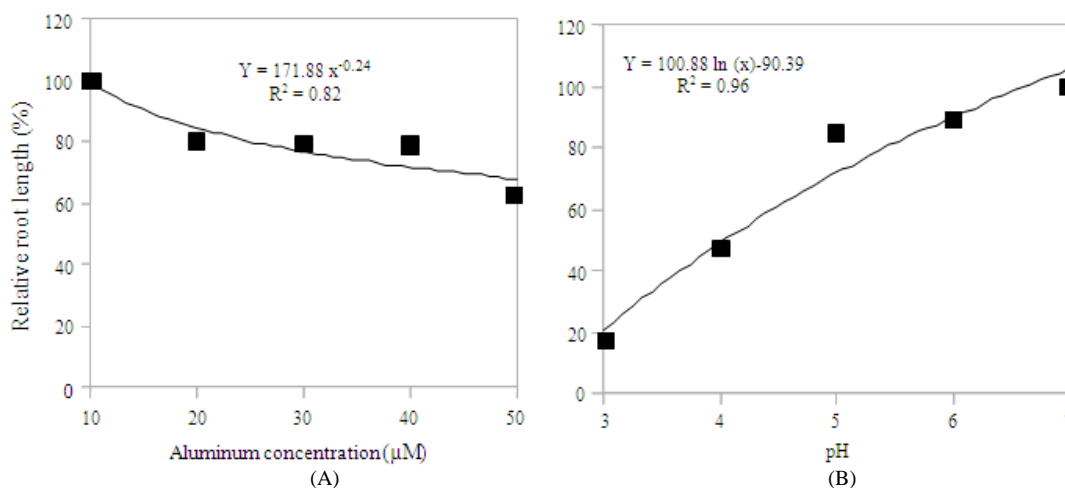


Fig. 3: The relationship between relative root length and Al (A) and pH (B)

Table 1: Exudation of organic acids by rice roots as affected by Al.

Aluminum concentration (µM)	Oxalate	Citrate	Malate
10	0.0008 <sup>a</sup>	0.0645 <sup>a</sup>	0.0077 <sup>a</sup>
20	0.0012 <sup>a</sup>	0.0005 <sup>a</sup>	0.0112 <sup>a</sup>
30	0.0019 <sup>a</sup>	0.0103 <sup>a</sup>	0.0215 <sup>a</sup>
40	0.0021 <sup>a</sup>	0.0051 <sup>a</sup>	0.0029 <sup>a</sup>
50	0.3100 <sup>a</sup>	0.0806 <sup>a</sup>	0.0766 <sup>a</sup>

Table 2: Exudation of organic acids by rice roots as affected by pH

pH	Oxalate	Citrate	Malate
3	0.0088 <sup>a</sup>	0.0066 <sup>a</sup>	0.0577 <sup>a</sup>
4	0.0006 <sup>a</sup>	0.0211 <sup>a</sup>	0.0005 <sup>a</sup>
5	0.0023 <sup>a</sup>	0.0168 <sup>a</sup>	0.0316 <sup>a</sup>
6	0.0028 <sup>a</sup>	0.0079 <sup>a</sup>	0.0072 <sup>a</sup>
7	0.0010 <sup>a</sup>	0.0061 <sup>a</sup>	0.0073 <sup>a</sup>

Means followed by the same letter within a column are not significantly different (HSD P<0.05)

We found that rice root released more organic acids at low pH than at high pH (Table 2). At the pH of 3, the seedlings released oxalate, citrate and malate at 0.0088, 0.0066-0.0577 µM, respectively.

## DISCUSSION

**Chemical properties of water from the field:** According to Chang *et al.* (1993), the area was

reclaimed from mangrove swamp in 1964 for rice cultivation. Up until now, the soils are very acidic (pH <3.5) even after many years of lime application. The average rice yield in the areas of acid sulfate soils in Malaysia using farmer's practice is less than 2 t/ha/season. The Al concentration is far above the critical toxic level of 74 µM for rice growth as suggested by Dent (1986). Hence, rice planted on a soil with this water condition is expected to suffer from H<sup>+</sup> stress (Zhu *et al.*, 2009). According to these authors, H<sup>+</sup> activity at pH 3.0 is more than 3000-fold higher than at pH 6.5. Rice grown on soils at Merbok would suffer from H<sup>+</sup> stress if the soils are not ameliorated with amendments. Hence, rice grown in the area would suffer from Al toxicity (Kochian *et al.*, 2005) and eventually the yield will be reduced.

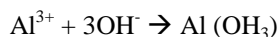
**Aluminum toxicity symptoms in plants:** Barker and Pilbeam (2007) stated that there were several symptoms of Al toxicity that occur within a few hours of Al exposure, such as inhibition of root elongation, disruption of root cap processes, lignin deposition and decline in cell division. This is consistent with the finding of Horst *et al.* (2009) who said that Al inhibited plant root elongation. It is believed that reduced root length results in decreased nutrient uptake and multiple nutrient deficiencies. Ridolfi and Garrec (2000) reported that the concentration of calcium and magnesium in beech leaf was reduced by Tan and Keljens (1995). Furthermore, Ryan *et al.* (1993) reported that roots of an aluminum-sensitive corn cultivar BR 201F was inhibited after 30 min of exposure to aluminum. This is because Al affects the growth of cells in the elongation zone of roots (Sasaki *et al.*, 1997).

The rice seedlings grow better under the condition of high pH than that of low pH. The pH of water in the area covered by acid sulfate soils in Malaysia is usually less than 3.5. That means rice growing on these soils is suffering from H<sup>+</sup> stress. This is consistent with the study of Zhu *et al.* (2009) who found that H<sup>+</sup> activity at low pH is much higher than at high pH. Hence, the root of rice growing in acid sulfate soils is affected by H<sup>+</sup> stress.

The Al concentration equivalent to 90% relative yield is 15 µM. This value is considered as the critical level of Al concentration, above which the growth of rice seedlings will be affected. It seems that variety MR 219 is less tolerant to Al toxicity than that reported by Dent (1986). Under normal circumstances, Al concentration of water in acid sulfate area cropped to rice (after lime is applied) is above that value. That is the reason why rice yield using farmer's practice is below the national average.

The critical pH value is 6. Under condition prevailing in acid sulfate soils, this level of pH can only be achieved by applying lime at high rate. Shamshuddin (2006) wrote that for these soils applying ground magnesium at 4 t/ha before rice planting only managed to raise the pH to about 4.5. Liming at higher rate than this is uneconomical and not sustainable in the long run. Again, it shows that growing rice on acid sulfate soils is difficult. Liming alone will not solve the problem of Al toxicity and H<sup>+</sup> stress in these soils. We have to use other means such as application of organic matter that can chelate Al to a certain extent.

The pKa value of Al is 5. So when the solution pH is raised above 5, Al will be precipitated as inert Al-hydroxides according to the following equation:



If this happens, rice growth stress would have been gradually diminishing, depending on how high the increase in pH is. We know for sure that when pH is raised to 5.2, Al concentration in the solution is minimal (Ismail *et al.*, 1993), no longer becoming a threat to the growth of crop like rice.

**Inhibition of root growth:** The decrease in root surface area due to high Al concentration can be similarly explained by deficiency of calcium and magnesium as reported by Ridolfi and Garrec (2000); Godbold *et al.* (1988) and Tan and Keltjens (1995). Arp and Strucel (1989) reported that water uptake of 1-year-old black spruce was reduced by exposure to high concentration of Al. Therefore, the presence of Al is really creating havoc in the growth of rice seedlings under the field conditions.

For better absorption of nutrients, the root surface area of rice seedling needs to be increased. In real situation in the field, this can be done by increasing solution pH using lime.

**Mechanism of Al tolerance:** Under real situation in the field in the vicinity of acid sulfate soils, water contains more than 50 µM Al even though the soils are limed at the standard and economic rate of 4 t ground magnesium limestone per hectare. There are several mechanisms of aluminum avoidance which include organic acid release, exudation of phosphate and alkalization of rhizosphere pH (Barker and Pilbeam, 2007). The phenomenon can be explained by lower concentration of Al at high pH because of its precipitation as gibbsite. In our case, as the pH increased, less organic acids were secreted by rice roots due immobilization of the phytotoxic Al<sup>3+</sup>. This is consistent with the explanation given by Kochian (1995).

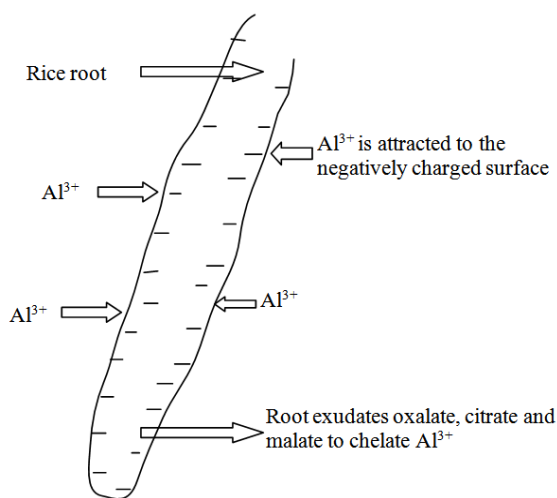


Fig. 4: Mechanism of aluminum tolerance in rice plant

In common bean, however, Al resistance is related to the sustained release of citrate from the root apices (Horst *et al.*, 2009). According to Ma *et al.* (2001),  $Al^{3+}$  was immobilized by organic acids that prevent it from entering the root cells. Under Al stress, it is known that root secretes citrate, forming Al-citrate that results in the reduction of Al toxicity. The same mechanism can be used to explain the slight tolerance of rice variety MR 219 to Al toxicity in this study. Detailed explanation of this mechanism is provided by Fig. 4. We think that organic acids had been exuded in order to detoxify the aluminum in the rhizosphere. The exudation of oxalic, citric and malic acids by rice root has been discussed earlier (Table 1-2). When these acids were released into the rhizospheres, they would effectively chelate  $Al^{3+}$  and prevented its entry into the root as explained by Lopez-Bucio *et al.* (2000). Furthermore, Ma *et al.* (2002) reported that the root apex of plant is particularly sensitive to Al and therefore, only the cations that immediately surround the apical root need to be detoxified.

As depicted in Fig. 4, the cell wall of rice root is negatively-charged. Hence, the positively-charged Al is attracted to the root surface. As soon as it touches the cell wall, the root takes immediate action by releasing oxalate, citrate and malate. These ligands, in turn, bind Al, forming Al-oxalate, Al-citrate or Al-malate. The result of these reactions is reduction of Al in the solution that causes toxicity to the rice seedlings. The amount of organic acids released differs from plant species to species. Miyasaka *et al.* (1991) found that aluminum-tolerant cultivar exuded many times more citrate than aluminum-sensitive cultivar in the presence

of aluminum and that oxalate appeared to be exuded less as compared to citrate. This is not consistent with the findings of this study where rice roots had released less citrate than oxalate.

## CONCLUSION

Rice variety MR 219 is affected by low pH and high Al concentration of the water in the paddy fields. If it is grown on acid sulfate soils in the country with water pH of  $< 3.5$  and Al concentration exceeding  $15 \mu M$ , the growth of rice will be curtailed and eventually the rice yield will be reduced. The water pH for optimal rice growth is 6. However, liming at the rate to achieve pH up to this level is costly, ordinary farmers cannot afford it. For this rice variety, in order to reduce the negative effects of  $Al^{3+}$  the root of the plant has to secrete organic acids, the most important of which is citric acid. The  $Al^{3+}$  in the solution will be partly chelated by the acid, forming Al-citrate, rendering it less available to the rice roots.

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